

Design and implementation of a new concept of photoelectrochemical cell for solar water splitting

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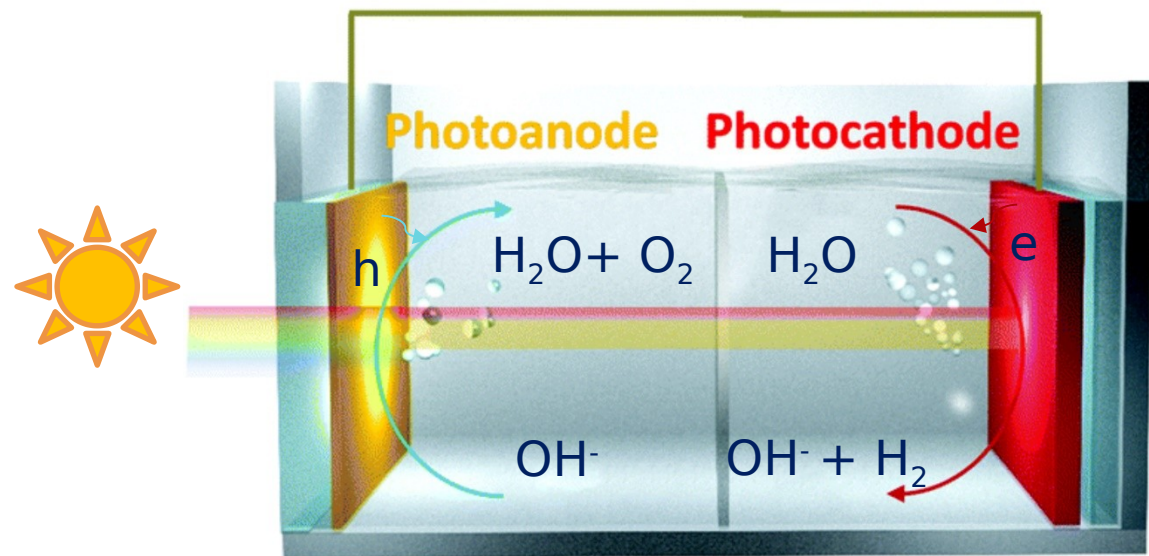


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Overview

- Photo-Electrochemical water splitting
- The FotoH2 concept
- Experimental results
- Scaled up device design
- Comparison vs test cell
- Summary

Photo-electrochemical water splitting: Basics



Anode:



Cathode:

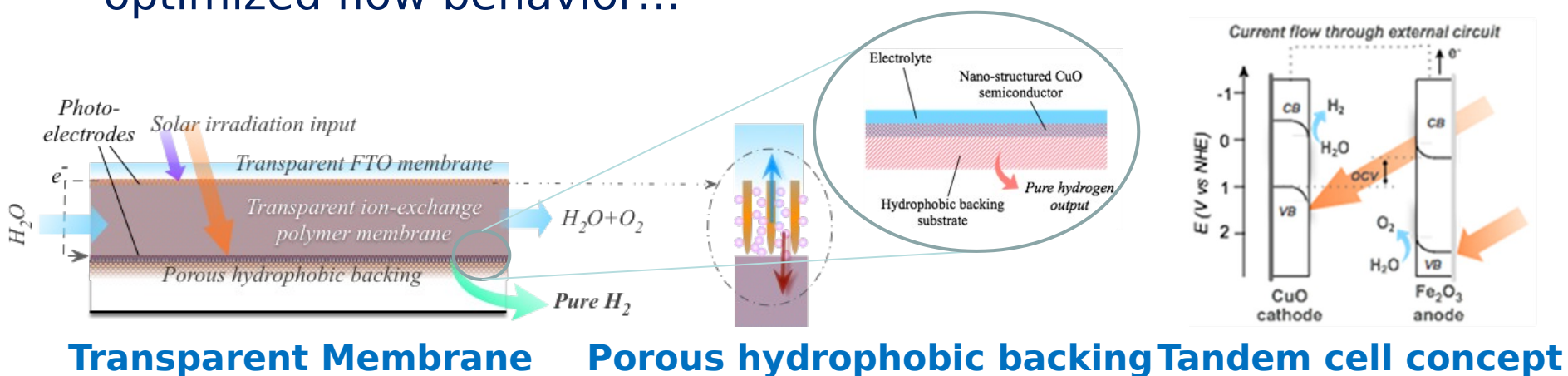


PEC cell : Photoelectrochemical cell for water splitting that converts solar energy directly to hydrogen fuel from photoelectrolysis. It comprises a photoelectrode and a counter electrode in contact with an electrolyte.

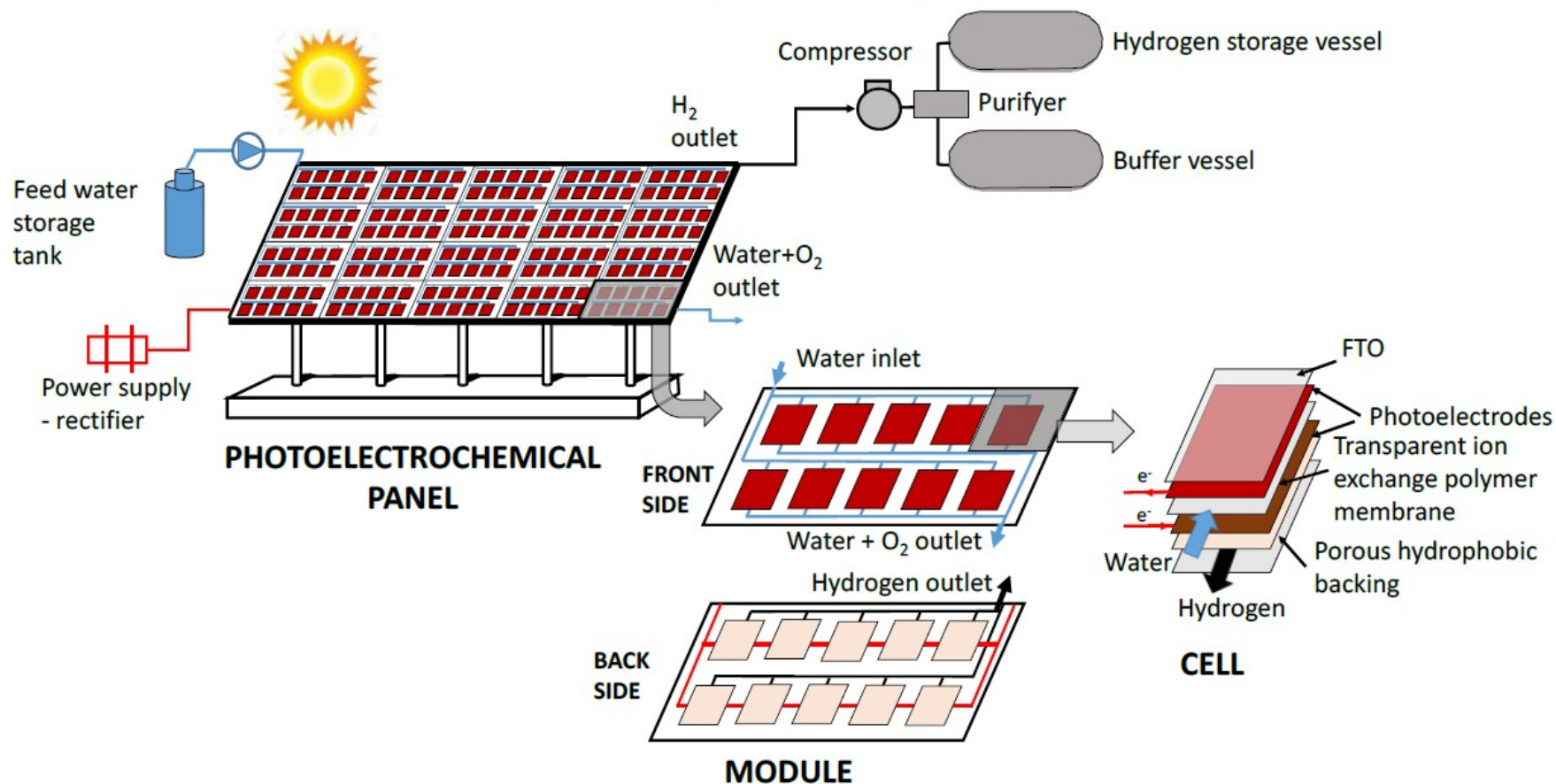
Photoelectrode: Photoelectrode is an electrode capable of initiating electrochemical transformations after absorbing light. In the solar water splitting context, photoelectrode is made of a semiconductor film on an electrically conductive material

The FotoH2 Concept

- Cheap photoelectrodes: Hematite (Fe_2O_3) and CuO
- Simple flow-cell design without corrosive electrolytes.
- Anion-exchange polymer membrane and porous hydrophobic backing concepts (solid electrolyte)
- Design of a tandem photo-electrochemical cell
- Large-scale prototyping and field testing
- Efficient low-cost photochemical water splitting reactors with optimized flow behavior...

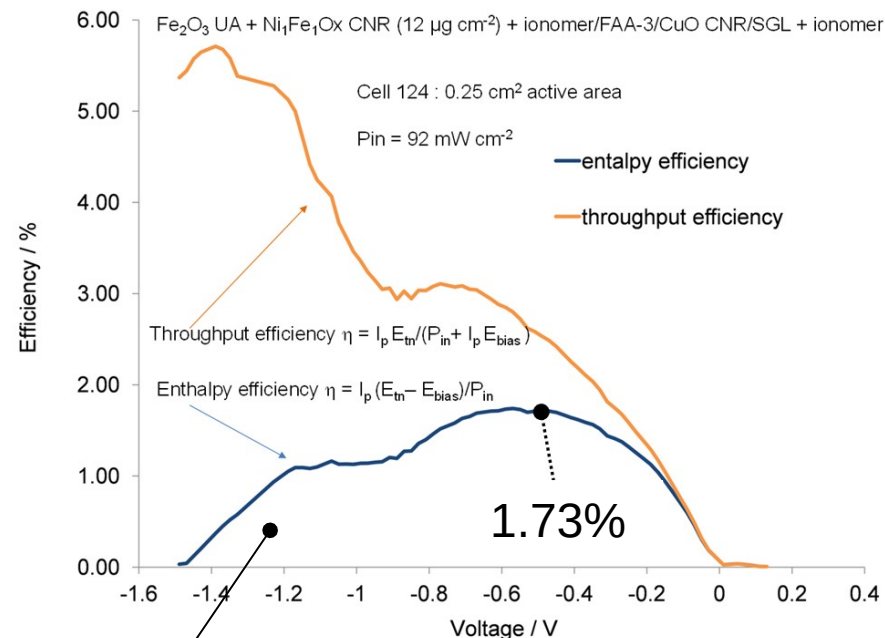
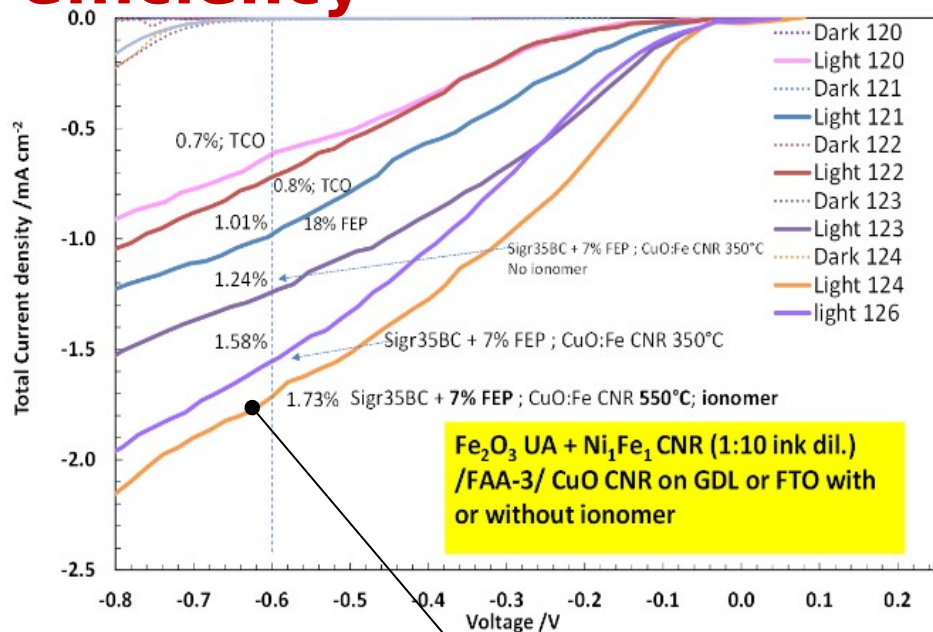


The FotoH2 Concept



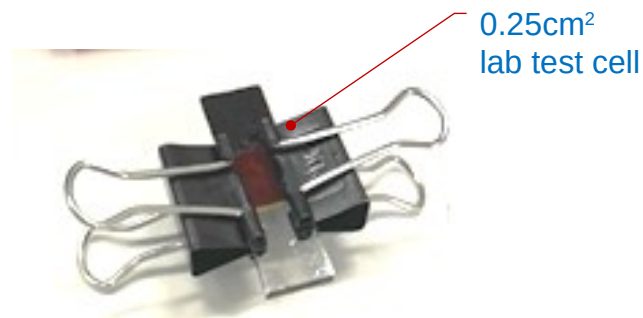
Modular approach as in Photo-Voltaic systems

Lab Test cell results: Polarization and efficiency

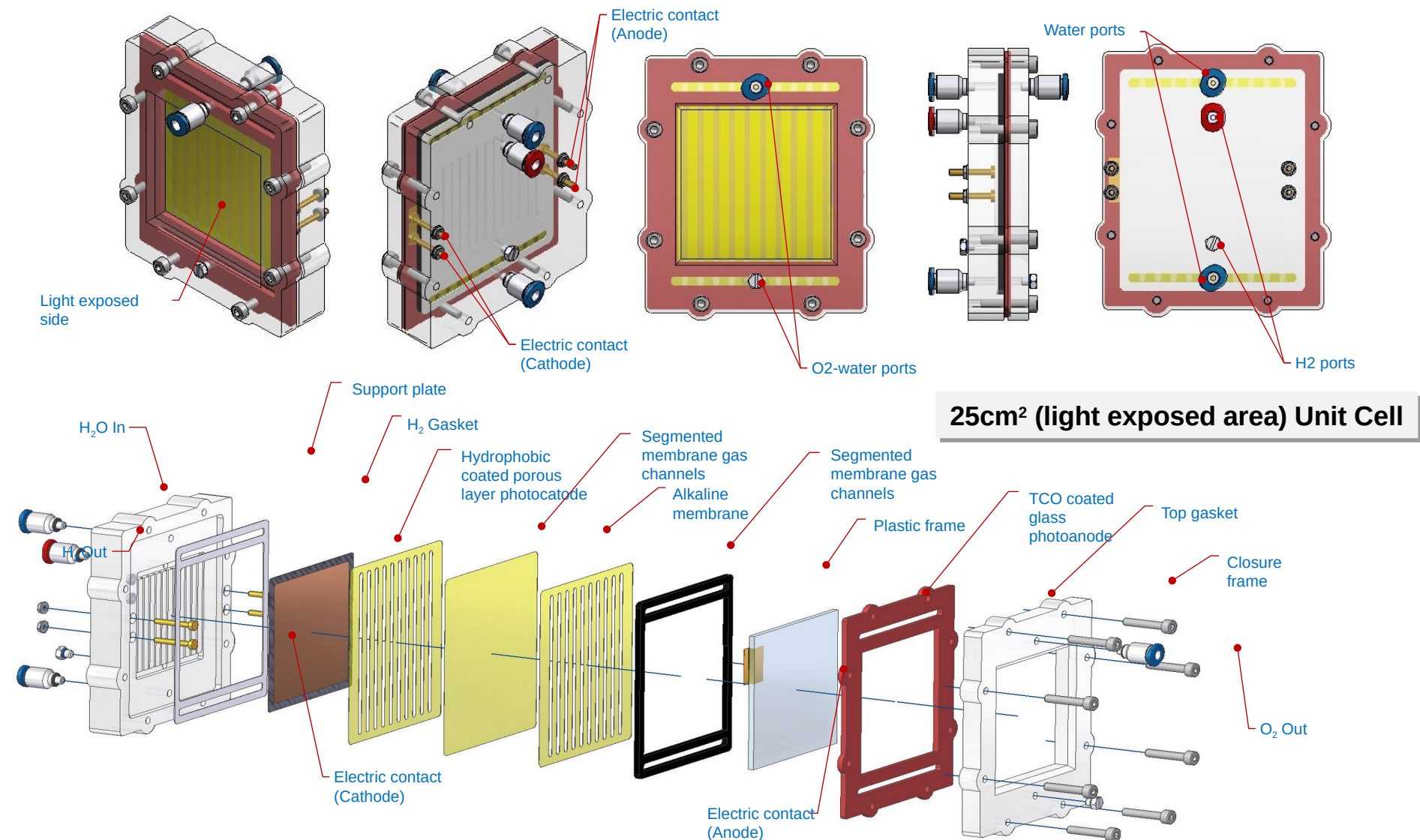


Electrode formulation screening

Efficiency calculation

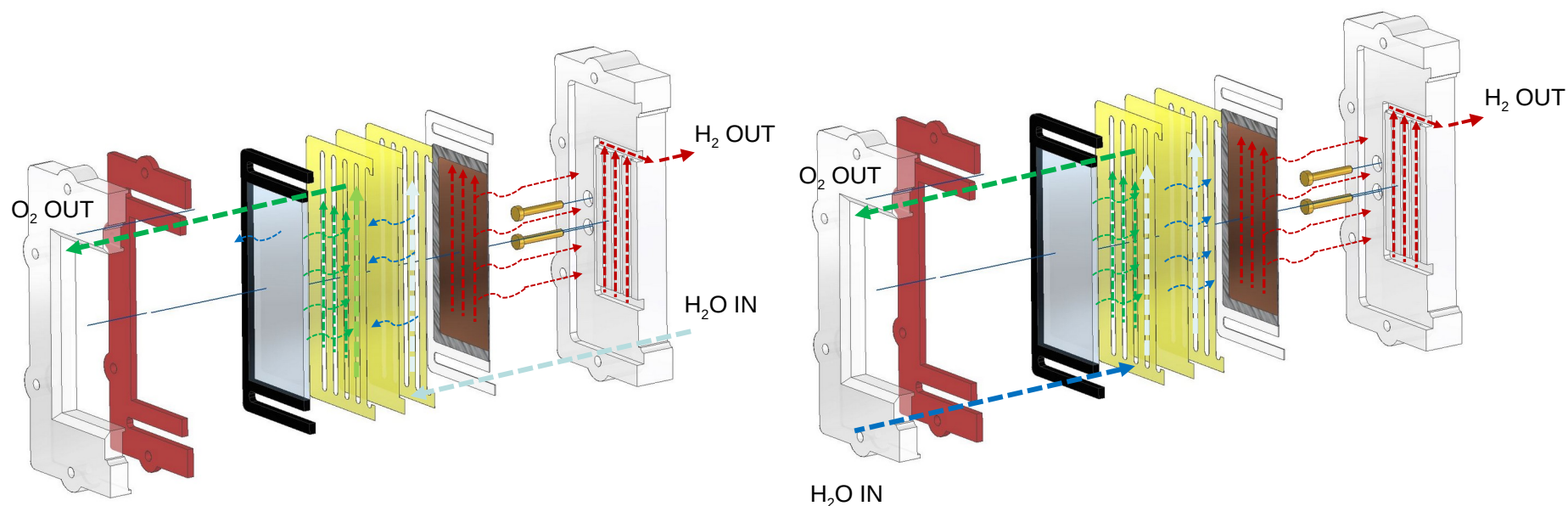


Scaled up cell Design



Reactants/Product Flow path

- Unit cell can operate by feeding water from anode and cathode respectively

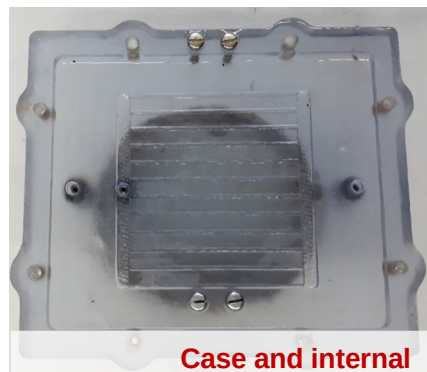


Water fed from the Cathode

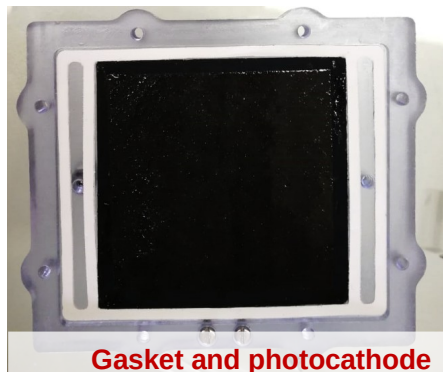
Water fed from the Anode

■ O₂
■ H₂
■ H₂O

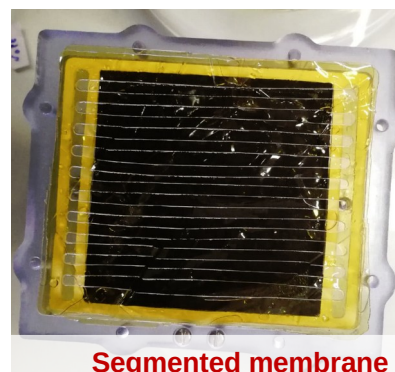
Unit cell components and assembly sequence



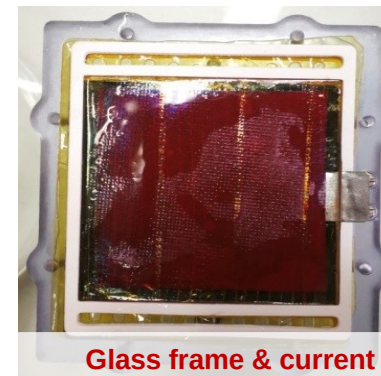
Case and internal connectors



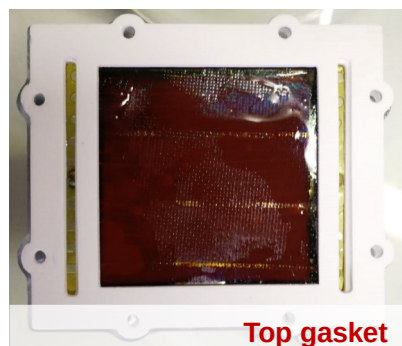
Gasket and photocathode



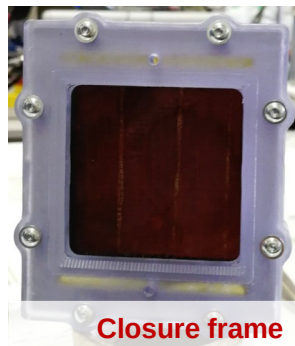
Segmented membrane



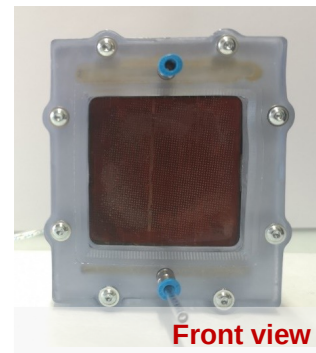
Glass frame & current slab



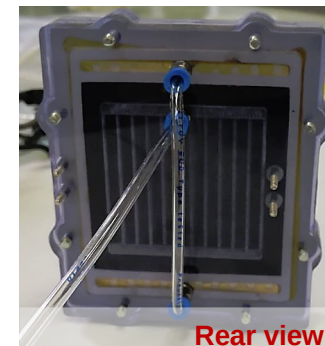
Top gasket



Closure frame



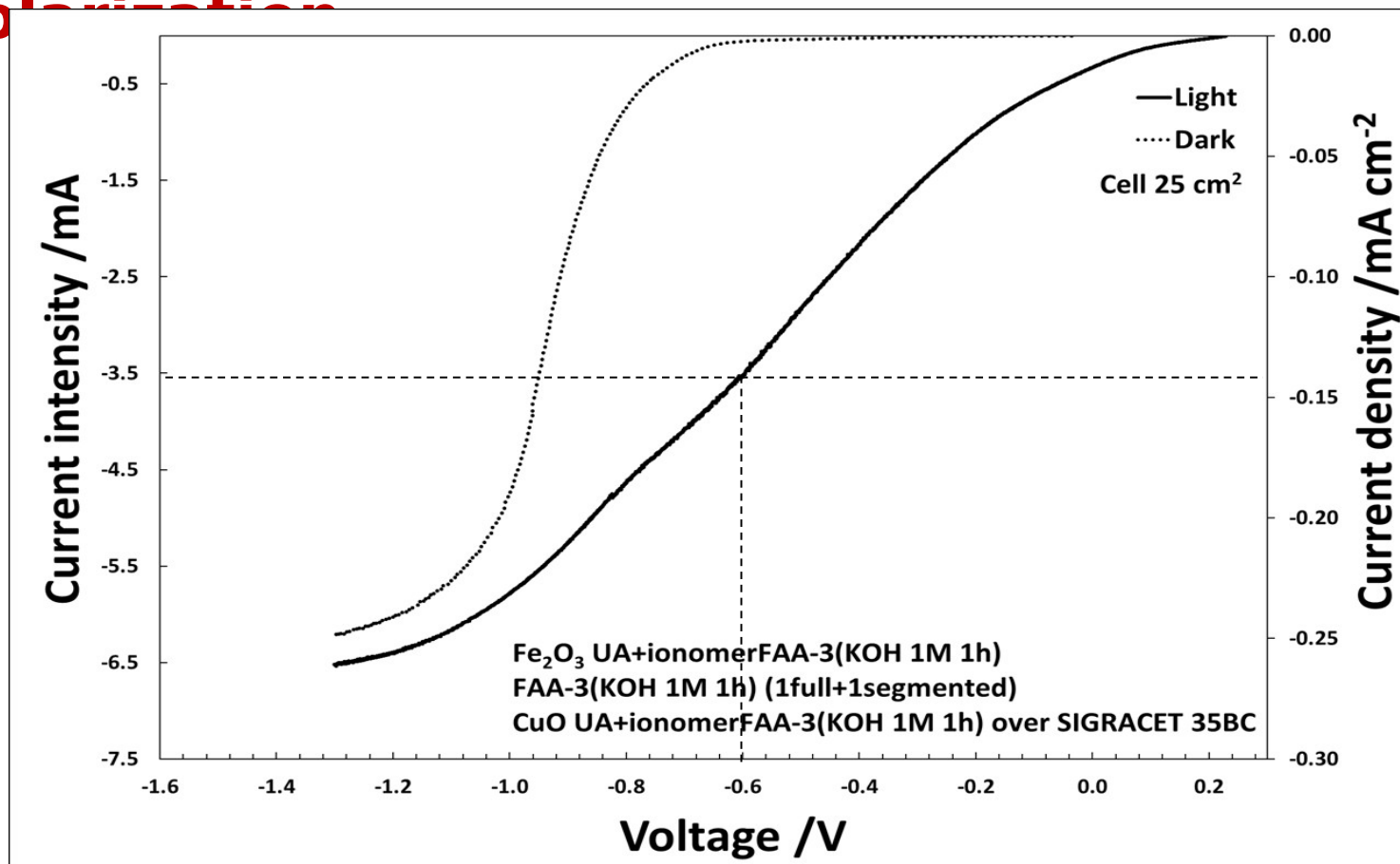
Front view



Rear view

PhotoElectrochemical tests scaled up:

Polarization

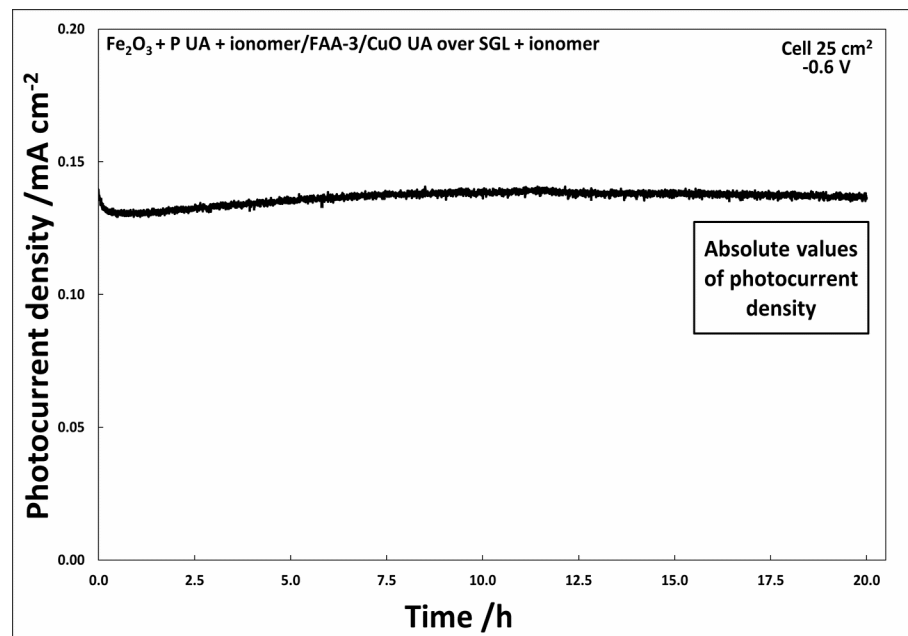
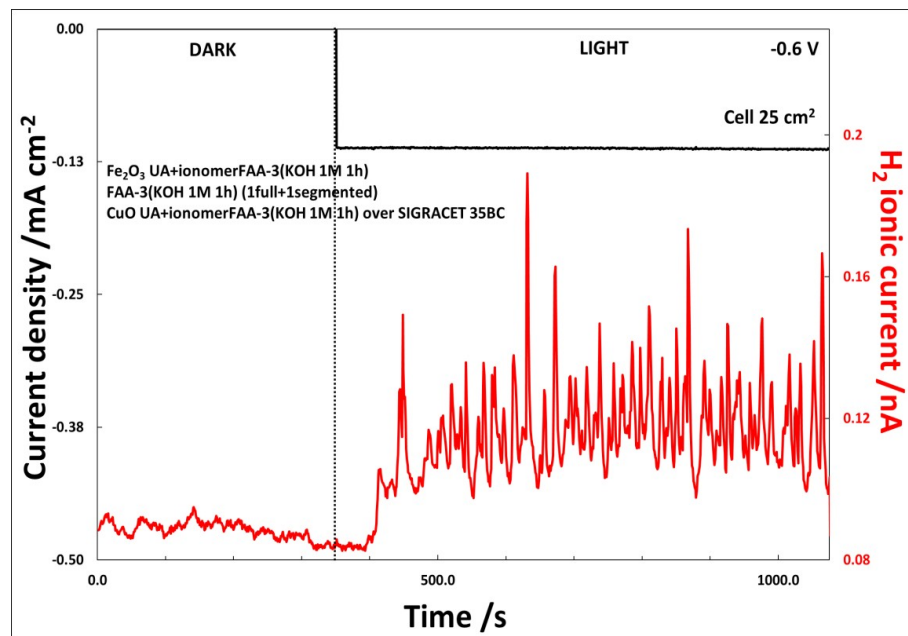


The photocurrent, in this case, is about **10 times lower** than the lab cell

PhotoElectrochemical tests : scaled up cell

Stability test

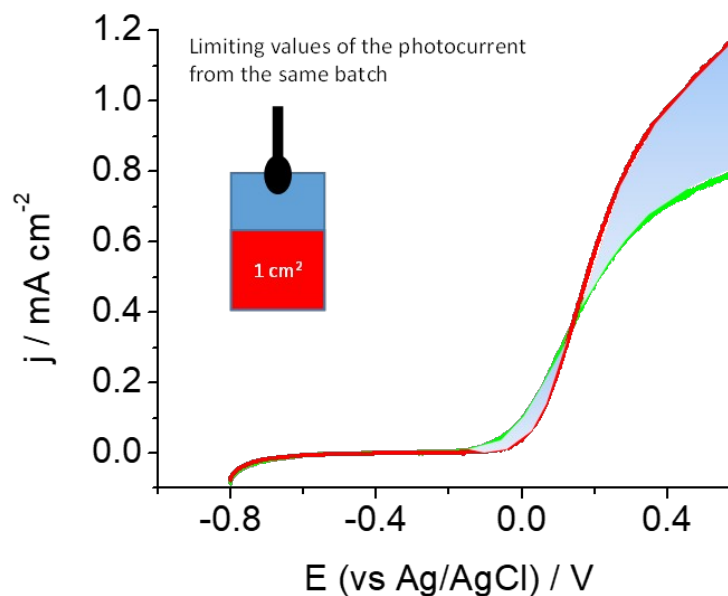
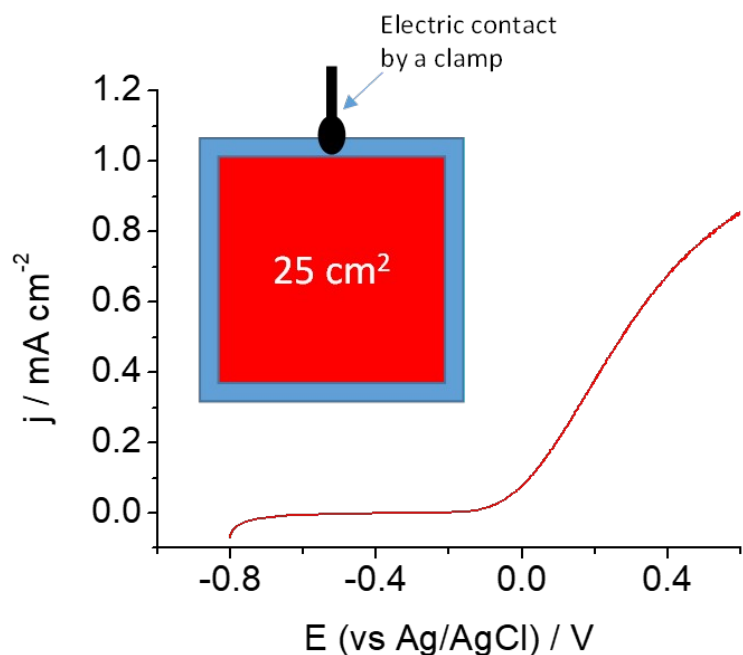
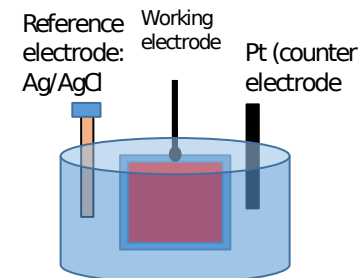
the photocurrent degradation is not appreciable.



Mass spectrometer measurements
 at the cathode output revealed
 numerous spikes related to hydrogen
 generation.

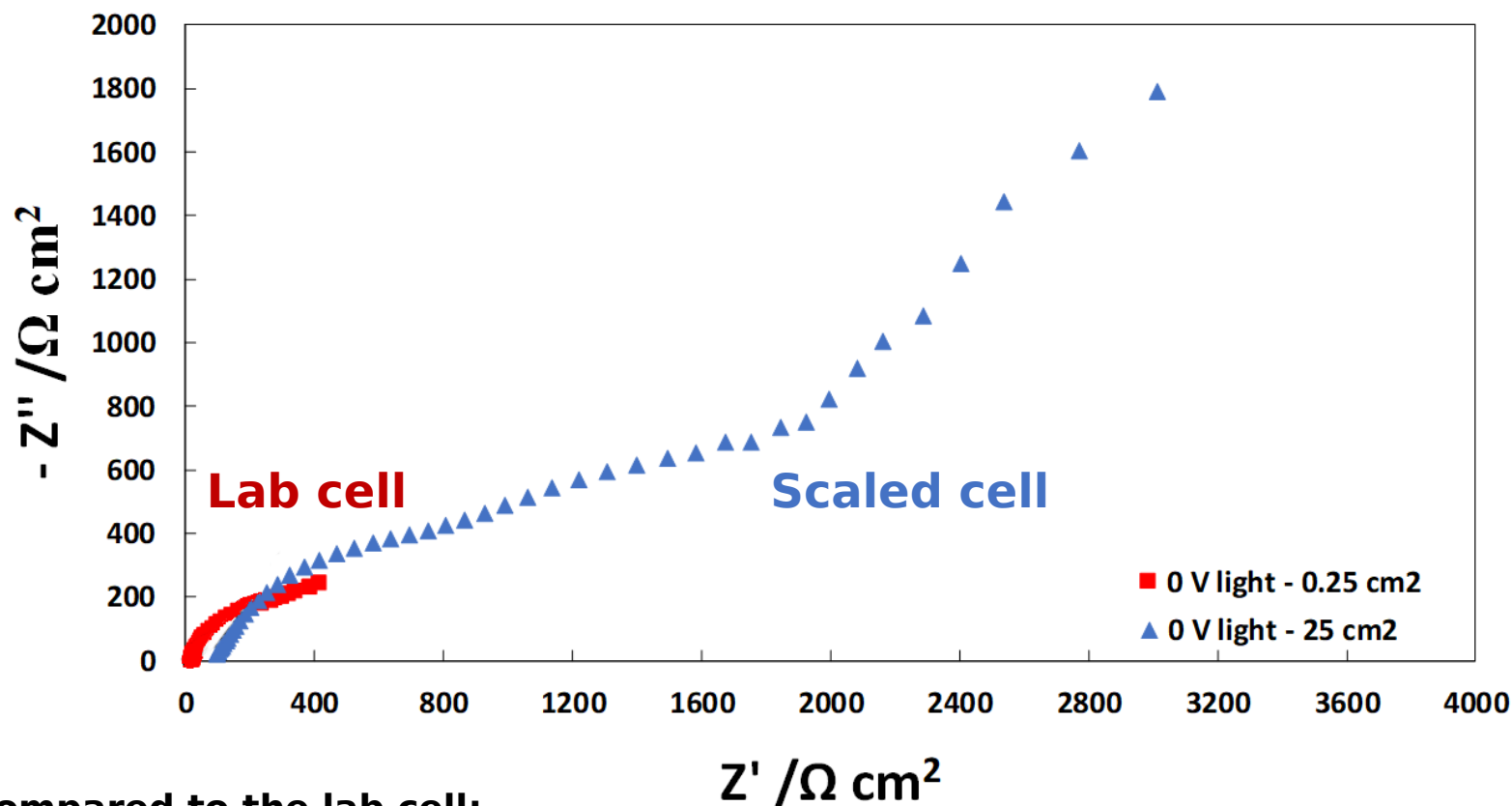
Large area photoelectrode response: **ex situ** test

Effect of the FTO Conductivity in the 25cm² hematite Photoanode



Conductivity of the glass appears not being the main limitation of the photocurrent

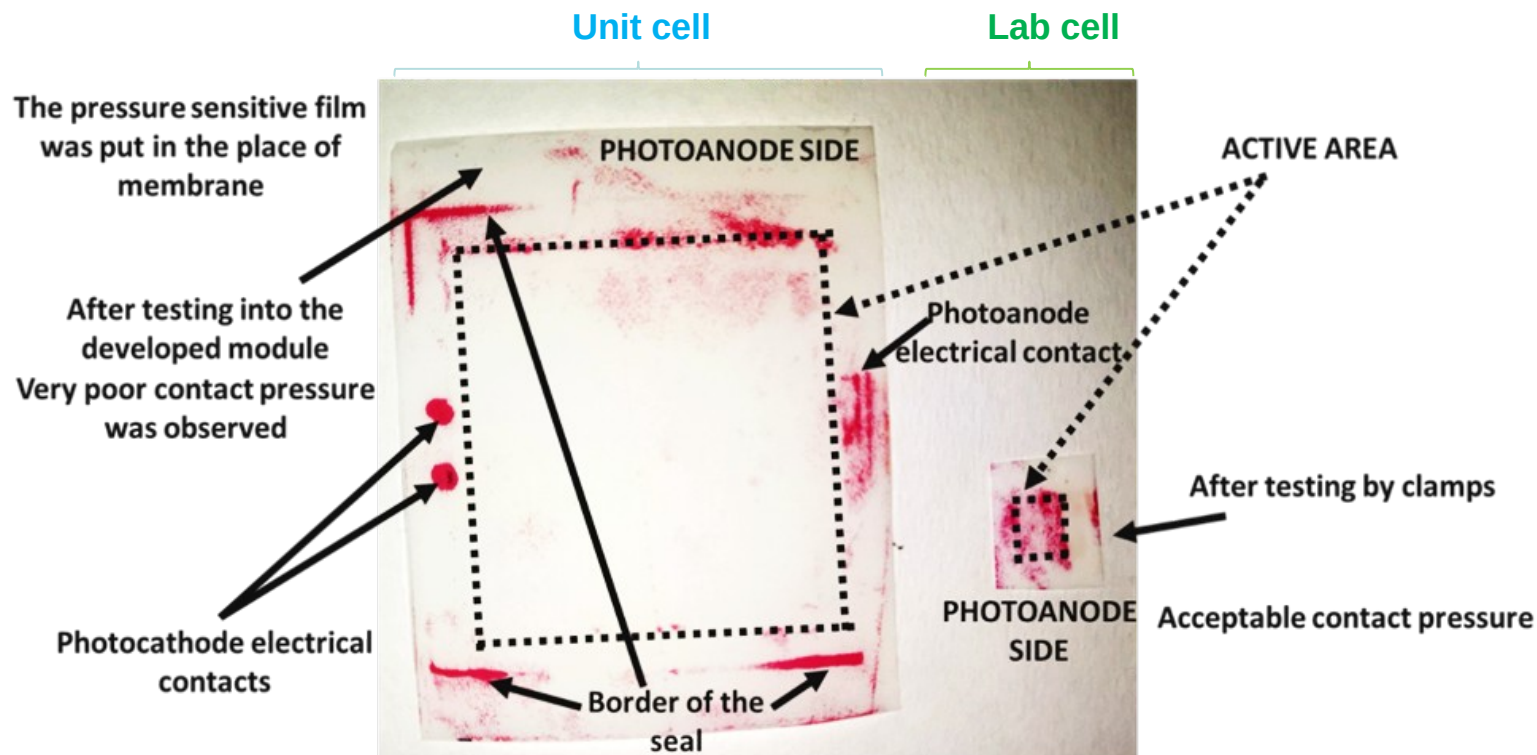
Unit Cell issues: EIS measurement



Compared to the lab cell:

- **High polarization resistance (R_p)**
- **High Series Resistance (R_s)**
- Main issues appear to be related to mass transfer at the photo electrode – electrolyte
- Recombination effect into the photoelectrode?

Contact pressure measurement



Contact pressure in the Unit Cell is lower than in the lab cell:

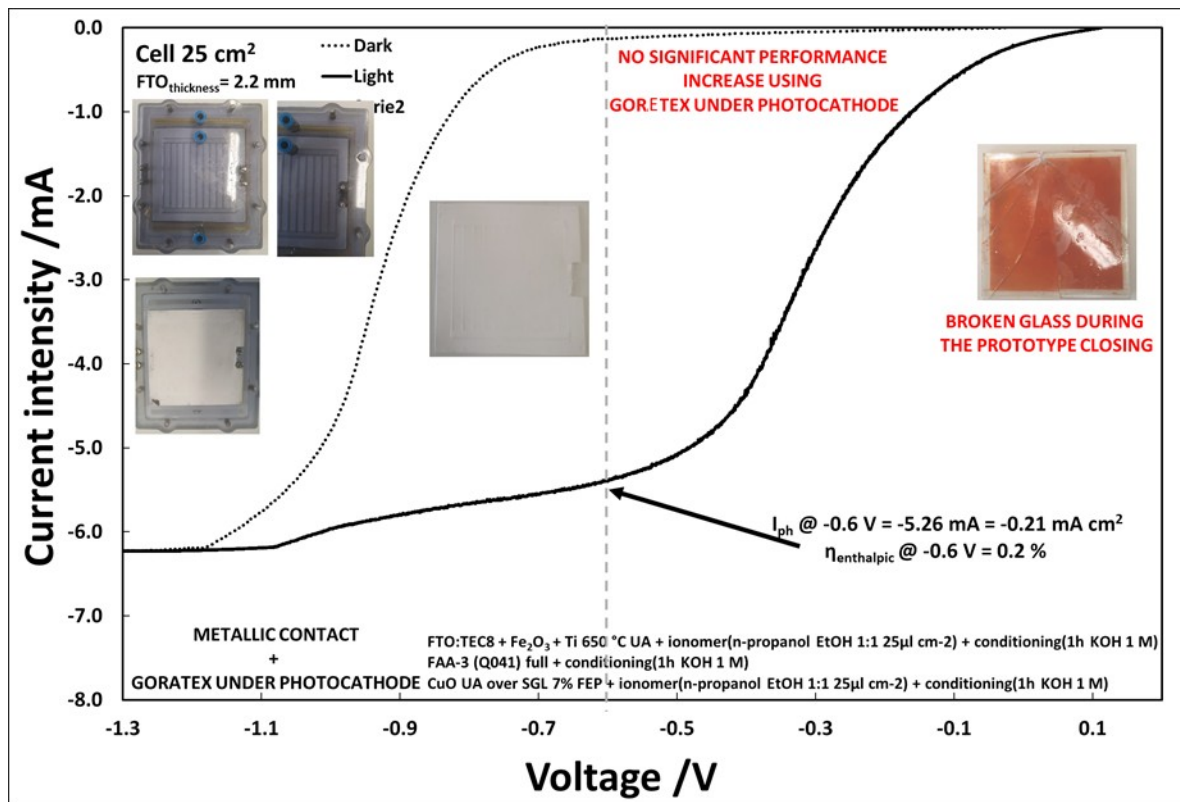
Possible Reasons

- Low clamping pressure applied
- Manufacturing tolerances inappropriateness
- Low closure frame stiffness

Early solutions

- Increase the clamping pressure
- Increase manufacturing tolerances i.e. planarity
- Stiffer material or design for closure frame.

Increasing the contact area

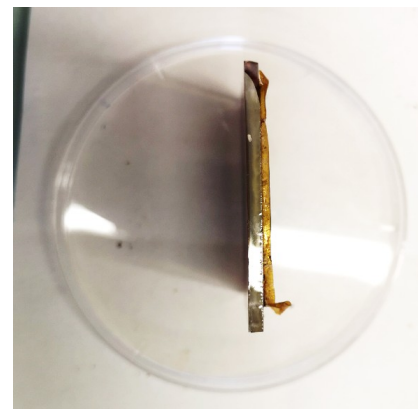
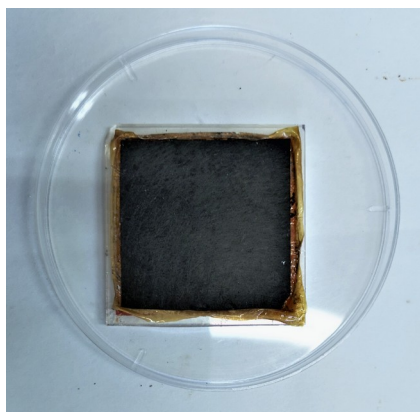


- Slight improvement in the photocurrent were recorded
- Mechanical issue due to differential displacements induced by Goretex

Approach to solve the present issues

- Improving the interface between photoelectrodes and membrane by ex situ assembly

- Hot pressing**
- Cold pressing “gluing process”**



- There are limiting factors in both approaches further work in needed.

Summary

A new concept of a PEC cell was demonstrated

The FotoH₂ cell showed a solar to Hydrogen conversion of **1.74%**

A scale up to 25cm² cell was proposed and implemented

Assembly issues required the hematite syntesys on FTO to be adjusted.

Electrochemical tests showed lower efficiency than the test cell.

Causes of such lower performance have been identified

New concept of cell assembly is under investigation.

Acknowledgements



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Thanks for your kind attention

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